1/4

Magnetic Materials

Recall that **atoms and molecules**, having both positive (i.e., protons) and negative (i.e., electron) charged particles can form **electric dipoles**.

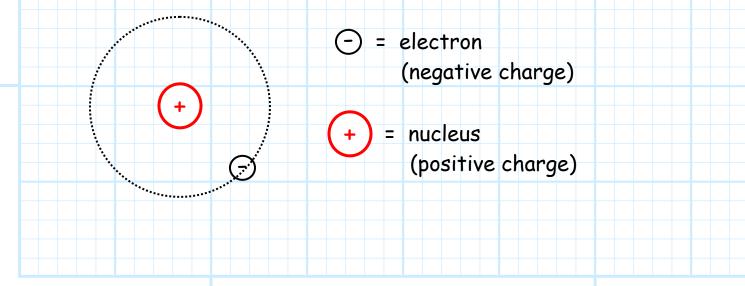
We find that atoms and molecules also can also form **magnetic dipoles**!

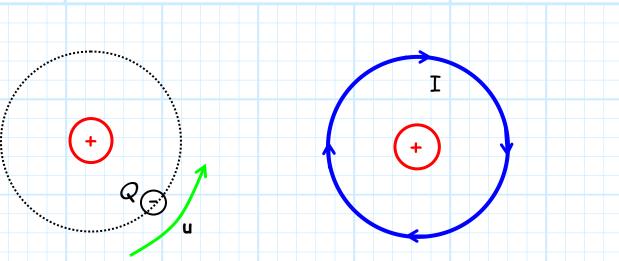
Q: How??

A: Recall a magnetic dipole is formed when current flows in a small loop. Current, of course, is moving charge, therefore charge moving around a small loop forms a magnetic dipole.

Molecules and atoms **often** exhibit electrons moving around in small loops!

Again, we use our **ridiculously** simple model of an atom:





An electron with charge Q orbiting around a nucleus at velocity **u** forms a small current loop, where $I = Q|\mathbf{u}|$.

This forms a magnetic dipole!

This is a **very simple** atomic explanation of how magnetic dipoles are formed in material. In actuality, the physical mechanisms that lead to magnetic dipoles can be **far** more complex. For example, **electron spin** can also create a magnetic dipole moment.

Typically, the atoms/molecules of materials exhibit either no magnetic dipole moment (i.e., $\mathbf{m} = 0$), or the dipole moments of each atom/molecule are randomly oriented, such that the net dipole moment is zero.

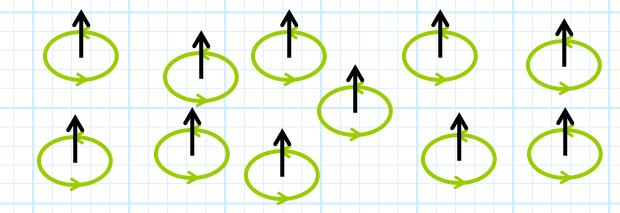
Therefore, if we have Nrandomly oriented magnetic dipoles \mathbf{m}_n , we find there average value will be zero:

$$\frac{1}{N}\sum_{n}\mathbf{m}_{n}=0$$

Similarly, we find that the **total** magnetic flux density created by these magnetic dipoles is **also zero**:

$$\sum_{n} \mathbf{B}_{n}(\bar{r}) = \mathbf{C}$$

However, we find that sometimes the magnetic dipole moment of each atom/molecule is **not** randomly oriented, but in fact are **aligned**!



In this case, total magnetic flux density created by these dipoles is **non-zero**!

$$\sum_{n} \mathbf{B}_{n}(\bar{\boldsymbol{r}}) \neq \mathbf{0}.$$



- A: Two possible reasons:
 - 1) the material is a permanent magnet.
 - 2) the material is immersed in some magnetizing field $B_m(\bar{r})$.